Organic & Biomolecular Chemistry

Cite this: Org. Biomol. Chem., 2012, 10, 6665

PAPER www.rsc.org/obc

Cycloisomerization of dienes and envnes catalysed by a modified ruthenium carbene species†

Álvaro Mallagaray, Kazem Mohammadiannejad-Abbasabadi,‡ Sandra Medina, Gema Domínguez and Javier Pérez-Castells*

Received 28th December 2011, Accepted 19th June 2012 DOI: 10.1039/c2ob07185a

Cycloisomerization is a totally atom economic procedure which converts dienes and enynes into cyclic molecules. Modification of Grubbs' 2nd generation catalysts by reaction with dimethylformamide provides a new species able to catalyse this transformation. Selection of suitable conditions allowed high yields and selectivity. Studies performed in order to identify the catalytic species point to a non-carbenic ruthenium complex that has lost the phosphine. No hydride signals appeared. In addition, the reaction works with enynes and the new species catalyses efficiently crossed cyclotrimerizations of alkynes with divnes.

Introduction

Non-metathetic transformations¹ are side reactions observed frequently when developing metathesis reactions catalysed by ruthenium alkylidene catalysts.² However, these alternative reactions, if optimized, may be useful transformations. Examples include oxidations, hydrosilylations of alkynes, hydrogenation of olefins, cyclopropanations, cycloaddition reactions and olefin isomerizations. We have shown recently the ability of second generation Grubbs' catalyst to mediate in a concurrent tandem catalysed triple process including RCM-isomerization and cyclopropanation.3 This transformation is achieved by heating the reaction after the completion of the RCM. Upon heating, the catalyst is presumably transformed into another species, possibly a ruthenium hydride able to produce a shift of the double bond formed in the RCM and to mediate in the final cyclopropanation step with ethyl diazoacetate. Many times the key issue in directing a particular process either to a metathesis or to a different transformation is to modify the ruthenium species before or during the reaction. Thus, various groups have reported on thermal transformations of [Ru]-I and [Ru]-II leading to species able to produce olefin isomerization.4 In addition, ruthenium hydrides can be obtained from [Ru]-II by reaction with methoxide or other additives. These hydrides mediate in double bond shifts,⁵ reductions⁶ and hydrovinylations.⁷ During our studies devoted to identify the catalytic species and the mechanism of the tandem RCM-isomerization-cyclopropanation transformation, we have found that when heated in dimethylformamide (DMF), [Ru]-II can be transformed into a new species that produces cycloisomerization of dienes. Recently, Arisawa and Nishida developed the selective cycloisomerization of dienes by a combination of [Ru]-II and vinyloxytrimethylsilane. They identified hydride A as the catalytic species and applied this methodology to the synthesis of indoles (Fig. 1).8

Cycloisomerizations can be catalyzed by various transition metals, such as palladium, nickel, rhodium and some ruthenium complexes like [RuCl₂(p-cymene)]₂ or [Ru(cod)Cl₂]₂. 11 Development of convenient selective methodologies based on efficient and easy to handle catalysts for the cycloisomerization of dienes is highly desirable and herein we present our results in

Fig. 1 Catalytic ruthenium species.

Universidad San Pablo-CEU, Departamento de Química, Facultad de Farmacia, Urb. Montepríncipe, 28668 Boadilla del Monte (Madrid), Spain. E-mail: jpercas@ceu.es

†Electronic supplementary information (ESI) available: Kinetics experiments, synthesis and isolation of modified carbene species and NMR spectra for new compounds. See DOI: 10.1039/c2ob07185a ‡ Present address: Catalysis Division, Department of Chemistry, University of Isfahan, Isfahan 81746-73441, Iran.

the use of ruthenium species obtained by reaction of Grubb's catalysts and DMF in cycloisomerization reactions.

Results and discussion

Our initial aim was exploring different reaction conditions with diene 1a (Table 1). Searching for RCM-isomerization products we explored the use of DMF as solvent. When performing a reaction at r.t., the metathesis product 3a was isolated in 89% yield as the only reaction product detectable in the ¹H NMR spectrum (entry 1). Heating the reaction to 120 °C, after mixing all the components, provided a mixture of 3a, its isomer 4a and the cycloisomerization product 2a (entry 2). In view of this result we decided to heat [Ru]-II for a short period of time and then add the diene continuing the reaction for 3 additional hours. Under these conditions (entry 3) conversion was low, but the major product was 2a albeit in low yield (17%) jointly with 5% of 3a. The finding of optimum conditions to obtain 2a was achieved with a subtle combination of time for the modification of the catalyst, catalyst loading and concentration. Finally, the best conditions were those of entry 6, with 10 mol% of catalyst, 10 min. for the modification of the complex and a concentration of 0.27 mM. Under these conditions total conversion was achieved and a 72% of cycloisomerization product 2a was isolated along with 7% of RCM product 3a. A slight variation of these conditions (entry 7) allowed total selectivity towards 2a (62% isolated yield) and a lower catalyst loading although without total conversion.

The ability of [Ru]-I and [Ru]-III to catalyse this reaction was checked next. As expected, due to its thermal instability [Ru]-I did not produce any identifiable product but an extensive decomposition of starting material (entry 10). On the other hand, [Ru]-III gave a mixture of RCM and RCM-isomerization products along with a small amount of cycloisomerization (entry 11). In an attempt to improve further the yields of this reaction avoiding the presumable decomposition of starting materials due to the high temperatures and with the aim to decrease the amount of DMF needed, we studied the possibility of carrying out the second part of the reaction in toluene. Thus, after modifying the complex in DMF (5 mL per mmol) a solution of the diene in toluene was added, reaching the optimized concentration, and reacted at reflux temperature (entry 12). To our delight the yield of 2a became excellent under these conditions (91%) and the reaction was completed after 30 min.

Once the feasibility of performing a selective cyloisomerization was proved, we studied the scope of the reaction with different 1,5-dienes using the conditions of entry 12 in Table 1. Results are summarized in Table 2.

The reaction gave excellent results with dienes 1b-d (entries 1-3). Protected amino groups were tolerated, giving tosyl derivative 1e good yields of cycloisomerization product 2e with a small amount of its thermodynamically stable isomer 2e' (90% yield jointly, entry 4). However, the parent BOC derivative 1f produced only 36% yield of 2f, and a double bond isomerization reaction was the major process and a (7:3) mixture of 5f + 5f'was isolated in 40% yield (entry 5). The cycloisomerization of amide 1g gave a mixture of the two possible lactams 2g and 2g' in good global yield (57%, entry 6). On the other hand, a double bond shift was the only process observed with sulfone 1h (entry 7).

The extension of the reaction to larger rings seem to be precluded as compound 6 gave only a 38% of a isomerizationcycloisomerization 5 membered-ring product 9. In addition, we

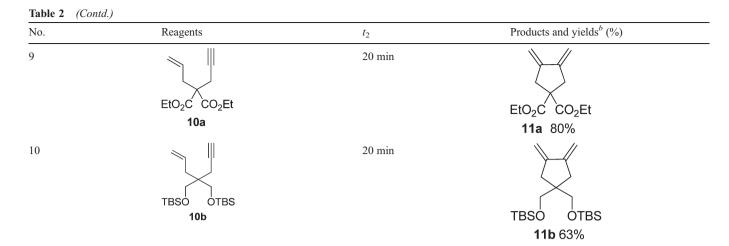
Table 1 Study of the cycloisomerization conditions^a

Entry	Cat. (mol%)	Conc 1a (mM)	t_1 (min)	<i>t</i> ₂ (h)	Conv.	Yield ^b (%)		
						2a	3a	4a
1	5 [Ru]-II	0.10	3 h at r.t.		100	n.d.	89	n.d.
2	10 [Ru]-II	0.10	3 h at 120 °C	2	100	10	50	16
3	5 [Ru]-II	0.10	3	3	15	17	5	n.d.
4	5 [Ru]-II	0.27	30	3	37	26	n.d.	n.d.
5	10 [Ru]-II	0.15	5	3	100	40	15	32
6	10 [Ru]-II	0.27	10	3	100	72	7	n.d.
7	7.5 [Ru]-II	0.27	5	3	85	62	n.d.	n.d.
8	10 [Ru]- II	0.44	5	3	95	59	7	n.d.
9	10 [Ru]-II	0.36	7.5	3	100	57	15	6
10	10 [Ru]-I	0.27	7.5	3	100	n.d.	n.d.	n.d.
11	10 [Ru]-III	0.27	7.5	3	100	4	47	44
12	10 [Ru]-II	0.27	12	0.5(tol)	100	91	6	n.d.

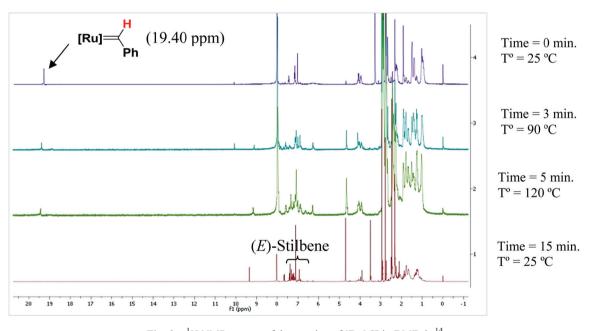
^a Reactions were carried out in anhydrous DMF (except entry 12, see text) at 120 °C except otherwise indicated and under inert atmosphere. ^b Yields in pure products. n.d. = not detected.

Table 2 Cycloisomerization reactions of dienes 1b-h, 6 and enynes 10a-b^a

No.	Reagents	t_2	Products and yields ^b (%)
1		16 h	
	BnO ₂ C CO ₂ Bn 1b		BnO ₂ C CO ₂ Bn
2		25 min	2b 93%
	1c		2c 66%
3		45 min	
	TBSÓ ÖTBS 1d		TBSO OTBS 2d 88%
4	N Ts	20 min	N N N N N N N N N N N N N N N N N N N
5	1e	1 h	†s †s †s †s 2e 71% 2e' 19% 5e 5%
	BOC 1f	161	BOC BOC BOC 2f 36% 5f + 5f' 40%
6	N O Dmob	16 h	N O N O Dmob
7	1g	3 h	2g 45% 2g' 12%
	S O ₂ 1h		S O ₂ 5h 90%
8		16 h	
	Dmob 6		N O N O N O N O N O N O N O N O N O N O



^a Reaction conditions: Ru-[II] (10 mol%) in DMF at 120 °C for 12 min then a solution of the substrate in toluene was added and refluxed. ^b Yields in pure products. Bn = benzyl; Dmob = 2,4-dimethoxybenzyl.



¹H NMR spectra of the reaction of [Ru]-II in DMF-d₇. ¹⁴

obtained mixtures of two isomers of the starting material, 7 and 8, that were separated and are formed upon one or two double bond shifts respectively. On the other hand, we studied the behavior of envnes 10a-b under the same reaction conditions. Interestingly, these substrates were transformed into 11a-b as the only reaction products (80 and 63% respectively). 12

Next we dedicated our efforts to try to identify the catalytic species formed upon the reaction of [Ru]-II with DMF. Thus, we performed ³¹P, ¹H and ¹³C NMR spectra of the ruthenium species before and after reaction with DMF and we also followed the transformation of the complex by ¹H NMR (Fig. 2). ¹³ From these spectra, ¹⁴ we observed the disappearance of the carbenic signal at 19.4 ppm as well as the signals corresponding to the phenyl group. At the same time (E)-stilbene appeared in the spectrum. After 15 min a new broad singlet at 9.18 ppm appeared corresponding to an imidazolinium salt. No ruthenium hydride species were detected. NMR measurements were conducted down to -35 ppm to check the possible presence of ruthenium hydrides which were not detected. In addition, the ³¹P spectra showed the loss of the ³¹P signal at 30 ppm while a signal at 50 ppm corresponding to P(O)Cy₃ appeared (see ESI†). From these data it can be presumed that the catalyst suffers successive loss of the phosphine and the benzylidene moieties. It is possible that the vacancies left are occupied by DMF molecules creating complexes similar to those described in the literature. 15 At the end of the process the species is totally decomposed and the only product isolated is an imidazolinium salt (9.18 ppm), ¹⁶ produced from the NHC ligand. The catalytic species could be an unstable complex formed along the process. Upon extraction of aliquots after 7 and 15 min of the catalyst modification

process we recorded sluggish ¹H- and ¹³C-NMR spectra (see Fig. 2 and ESI†) and the IR spectra showed absorptions around 1630 cm⁻¹ plus one absorption at 1947 cm⁻¹ which may correspond to a CO ligand bonded to ruthenium¹⁵ which would come from thermal decomposition of DMF into CO and dimethylamine. MS spectra at 7 min showed a peak at 1002 which suggests a dimeric nature of an intermediate complex. This hypothetical structure could be similar to the ones described by Grubbs in the studies on the thermal decomposition of [Ru]-II and other ruthenium-based carbenic catalysts. 17 If the mixture is analysed after 15 min it becomes more complex and we can see peaks at 1298, 1669 and 1882. Unfortunately, we were not able to isolate a pure sample of the active species. Attempts to separate it from imidazolinium salts phosphine oxides and other subproducts led to extensive decomposition. The inability of the new species to catalyse metathesis whereas it is capable to produce both cycloisomerization and double bond shifts means a transformation into one or more active catalysts possibly with a short life. The preference for each process seems to be related mainly to the structure of the substrate.

Following our recent results in the use of ruthenium carbenes in cyclotrimerization reactions of alkynes, 18 we checked the ability of the new species to promote this transformation. In Scheme 1 three crossed-cyclotrimerization reactions are shown between divnes 12a-b and phenylacetylene or 1-phenylpropyne. The reaction was highly efficient even when constructing sterically crowded products as 13c. As the cyclotrimerization reaction catalysed by ruthenium complexes is presumed to proceed through a cascade metathesis mechanism, this result opens the possibility of a different reaction pathway with this complex, possibly related with that observed with other metal complexes used in cyclotrimerization reactions.

Scheme 1 Crossed-ciclotrimerization reactions of 12a-b catalysed by the modified ruthenium species.

13c: R = R" = Me, R' = CO₂Bn, 71%

Conclusions

In summary, second generation Grubb's catalyst has been modified by reaction with DMF. We have shown the ability of the new species to catalyse several isomerization reactions with dienes, enynes and diynes. Extension of this methodology to more substrates and further studies on this topic are currently underway. Funding of this project by Spanish MEC (No. CTQ2009-07738/BQU) is acknowledged. A.M. and S.M. thank FUSP-CEU for a pre-doctoral fellowship.

Experimental section

General procedures

¹H NMR, ¹³C NMR and ³¹P NMR spectra were acquired on Bruker AM-300, Bruker AV400 and Bruker AVIII 700 spectometers. Chemical shifts^(TM) are in parts per million relative to tetramethylsilane at 0.00 ppm. IR spectra were determined by a FT-IR Perkin-Elmer 2000 spectrometer. TLC analyses were performed on commercial aluminium sheets bearing 0.25 mm layer of Silica gel. Silica gel 0.035-0.070 mm, 60 Å was used for column chromatography. Anhydrous N,N-dimethylformamide (DMF) was purchased and used without further purification. Hexane and toluene were refluxed over calcium hydride. All reactions were conducted under argon atmosphere.

Preparation of starting materials

Products 1a, 19 1b, 19 1c, 20 1d, 20 1e, 21 1f, 21 1g, 3 1h, 22 6, 3 10a, 23 and 12a²⁴ were prepared according to the reported procedures. Compounds 3a, 25 4a, 26 $5e^{27}$ have already been described and matched with the bibliography data.

Preparation of 6-allyl-2,2,3,3,9,9,10,10-octamethyl-6-(prop-2-ynyl)-4,8-dioxa-3,9-disilaundecane (10b). 2-Allyl-2-(prop-2-ynyl)propane-1,3-diol²⁸ (400.0 mg, 2.6 mmol), TBSCl (1.70 g, 10.4 mmol) and imidazol (884.0 mg, 13.0 mmol) were dissolved in anhydrous DMF (13 mL) and the reaction was conducted under argon for 2 h at r.t. A mixture of water-ice (30 mL) and diethyl ether (30 mL) was added and the organic layer was extracted, washed with water (5 × 10 mL), brine $(3 \times 10 \text{ mL})$, dried with MgSO₄ and the solvent was evaporated. The resulting oil was purified by performing a silica gel chromatography (hexane, $R_f = 0.88$ in hexane) obtaining **10b** (862 mg, 2.25 mmol, 87%) as a colorless oil. ¹H NMR (300 MHz, CDCl₃) δ 5.84–5.60 (m, 1H, *CH*=CH₂), 5.08–5.01 (m, 2H, $CH = CH_2$), 3.41–3.34 (m, 4H, 2 × CH_2O), 2.08–2.06 (m, 4H, $2 \times CH_2$), 1.89 (t, J = 2.6 Hz, 1H, C=CH), 0.85 (s, 18H, $2 \times$ $C(CH_3)_3$), 0.00 (s, 12H, 2 × $(CH_3)_2Si$)); ¹³C NMR (75 MHz, CDCl₃) δ 134.2, 117.8, 81.7, 69.8, 63.6, 43.3, 35.1, 25.9, 21.0, 18.3, -5.6; IR (neat) 3077, 2955, 2930, 2858, 1640 cm⁻¹; Anal. Calcd for C₂₁H₄₂O₂Si₂: C, 65.90; H, 11.06. Found: C, 66.04; H, 11.18.

Preparation of dibenzyl 2,2-di(but-2-ynyl)malonate (12b).

Over a suspension of NaH (288.2 mg, 12.0 mmol) in anhydrous THF (20 mL) at 0 °C and under argon, a solution of dibenzyl malonate²⁹ (1.36 g, 4.80 mmol) in anhydrous THF (5 mL) was added and the mixture was heated to r.t. and stirred for 30 min. 1-Bromobut-2-yne (1.40 g, 10.6 mmol, 0.95 mL) was added and the reaction was stirred at r.t. until no more starting material was observed (TLC). Water (20 mL) was slowly added and the crude was extracted with DCM (3 × 40 mL). The organic layers were dried with MgSO₄, organic solvents were evaporated and the resulting oil was purified by performing a silica gel chromatography (Hex–AcOEt 9:1, $R_f = 0.57$ in Hex–AcOEt 4:1) obtaining 12b (1.27 g, 3.26 mmol, 68%) as a colorless oil. ¹H NMR (300 MHz, CDCl₃) δ 7.33–7.25 (m, 10H, 2 × Ar), 5.13 (s, 4H, $2 \times Ar-CH_2O$), 2.94 (d, J = 2.5 Hz, 4H, $2 \times CH_2C = CCH_3$), 1.67 (t, J = 2.5 Hz, 6H, $2 \times \text{CH}_2\text{C} = \text{C}CH_3$); ¹³C NMR

(75 MHz, CDCl₃) δ 169.0, 135.4, 128.5, 128.2, 128.1, 79.2, 73.1, 67.3, 57.3, 23.0, 3.5; IR (neat) 3034, 2957, 2921, 1740, 1607, 1587, 1498 cm⁻¹; Anal. Calcd for C₂₅H₂₄O₄: C, 77.30; H, 6.23. Found: C, 77,13; H, 6.16.

General procedure for cycloisomerization reaction

Catalyst [Ru]-II (36 mg, 0.042 mmol) was placed in a flamedried two-necked flask equipped with a condenser, and two cycles of vacuum-argon were performed. Anhydrous DMF (0.2 mL) was added and the suspension was heated at 120 °C for 12 min. The dark solution was cooled to r.t., anhydrous toluene (0.9 mL) was added followed by the addition of diene (0.42 mmol) in anhydrous toluene (1 mL). The mixture was gently refluxed until no more starting material was detected (TLC), cooled to r.t., filtered through Celite and solvents were removed under reduced pressure. The resulting dark-brown oil was purified by flash chromatography.

Diethyl 3-methyl-4-methylenecyclopentane-1,1-dicarboxylate (2a).²⁹ Following general procedure for cycloisomerization, starting with **1a** (100 mg, 0.416 mmol) after 25 min **2a** was obtained after silica gel chromatography (Hex–AcOEt 49:1, $R_f = 0.63$ in Hex–AcOEt 9:1) as a colorless oil (91 mg, 0.38 mmol, 91%). ¹H NMR (300 MHz, CDCl₃) δ 4.91 (d, J = 2.0 Hz, 1H, C=CHa), 4.80 (d, J = 2.1 Hz, 1H, C=CHb), 4.19 (q, J = 7.2 Hz, 2H, CO₂CH₂CH₃), 4.18 (q, J = 7.0 Hz, 2H, CO₂CH₂CH₃), 3.08–2.90 (m, 2H, CO₂CCH₂C=CH₂), 2.57–2.51 (m, 2H, CO₂CCH₂CH), 1.80–1.70 (m, 1H, CO₂CCH₂-CH), 1.25 (t, J = 7.0 Hz, 3H, CO₂CH₂CH₃), 1.11 (d, J = 6.1 Hz, 3H, CCH₃); Anal. Calcd for C₁₃H₂₀O₄: C, 64.98; H, 8.39. Found: C, 65.14; H, 8.47.

Dibenzyl 3-methyl-4-methylenecyclopentane-1,1-dicarboxylate (2b). Following general procedure for cycloisomerization, starting with **1b** (100 mg, 0.27 mmol) after 16 h **2b** was obtained after silica gel chromatography (Hex–AcOEt 20 : 1, $R_{\rm f}$ = 0.53 in Hex–AcOEt 9 : 1) as a colorless oil (93 mg, 0.25 mmol, 93%). ¹H NMR (300 MHz, CDCl₃) δ 7.31–7.22 (m, 10H, Ar), 5.11 (s, 2H, Ar–CH₂O), 5.11 (s, 2H, Ar–CH₂O), 4.90 (d, J = 2.1 Hz, 1H, C=CHa), 4.79 (d, J = 2.2 Hz, 1H, C=CHb), 3.11–2.93 (m, 2H, CO₂CCH₂C=CH₂), 2.63–2.53 (m, 2H, CO₂CCH₂CH), 1.83–1.74 (m, 1H, CO₂CCH₂CH), 1.09 (d, J = 6.3 Hz, 3H, CCH₃); ¹³C NMR (75 MHz, CDCl₃) δ 172.0, 171.9, 153.5, 135.9, 128.9, 128.6, 128.3, 106.1, 67.6, 67.5, 58.8, 42.5, 41.0, 37.6, 18.4; IR (neat) 3067, 3034, 2961, 2932, 2873, 1733, 1659 cm⁻¹; Anal. Calcd for C₂₃H₂₄O₄: C, 75.80; H, 6.64. Found: C, 75.72; H, 6.54.

2,8,8-Trimethyl-3-methylidene-7,9-dioxaspiro[**4.5**]**decane (2c).**²⁰ Following general procedure for cycloisomerization, starting with **1c** (100 mg, 0.509 mmol) after 25 min **2c** was obtained after silica gel chromatography (Hex–AcOEt 49:1, $R_{\rm f}$ = 0.45 in Hex–AcOEt 9:1) as a colorless oil (66 mg, 0.33 mmol, 66%). ¹H NMR (300 MHz, CDCl₃) δ 4.86 (s, 1H, C= CH_2), 4.78 (s, 1H, C= CH_2), 3.64–3.55 (m, 4H, 2 × CH₂O), 2.54–2.46 (m, 1H, $CHCH_3$), 2.42–2.17 (m, 2H, $CH_2C=CH_2$), 1.99 (dd, 1H, J_1 = 13.0 Hz, J_2 = 8.2 Hz, CH_2CHCH_3), 1.43 (s, 3H, CH₃C), 1.42 (s, 3H, CH₃C), 1.08 (d, J_1 = 6.7 Hz, 3H, CH_3CH_3), 1.03 (dd, J_1 =

12.8 Hz, $J_2 = 10.5$ Hz, 1H, CH_2 CHCH₃); Anal. Calcd for $C_{12}H_{20}O_2$: C, 73.43; H, 10.27. Found: C, 73.61; H, 10.12.

(3-Methyl-4-methylenecyclopentane-1,1-diyl)bis(methylene)bis(oxy)bis(tert-butyldimethylsilane) (2d). Following general procedure for cycloisomerization, starting with 1d (100 mg, 0.26 mmol) after 45 min 2d was obtained after silica gel chromatography (hexane, $R_f = 0.50$ in hexane) as a colorless oil (88 mg, 0.23 mmol, 88%). H NMR (300 MHz, CDCl₃) δ ¹H NMR (300 MHz, CDCl₃) δ 4.77 (s, 1H, C= CH_2), 4.69 (s, 1H, C= CH_2), 3.41–3.33 (m, 4H, 2 × CH₂O), 2.49–2.45 (m, 1H, $CHCH_3$), 2.14 (s, 2H, $CH_2C=CH_2$), 1.79 (dd, 1H, $J_1 = 12.9$ Hz, $J_2 = 8.3$ Hz, CH_2CHCH_3), 1.08 (d, J = 6.7 Hz, 3H, CH_3CH), 0.97 (dd, $J_1 = 12.8$ Hz, $J_2 = 10.5$ Hz, 1H, CH_2CHCH_3), 0.86 (s, 18H, 2 × C(CH₃)₃), 0.00 (s, 12H, 2 × (CH₃)₂Si)); Anal. Calcd for $C_{21}H_{44}O_2Si_2$: C, 65.56; H, 11.53. Found: C, 65.69; H, 11.64.

3-Methyl-4-methylene-1-[(4-methylphenyl)sulfonyl]pyrrolidine (2e). ³⁰ Following general procedure for cycloisomerization, starting with 1e (100 mg, 0.398 mmol) after 20 min and purification through silica gel chromatography (Hex–AcOEt 20:1, $R_{\rm f}=0.29$ for 2e and $R_{\rm f}=0.24$ for 2e', both in Hex–AcOEt 9:1) 2e (71 mg, 0.282 mmol, 71%) was obtained as a white solid (Mp: 60–63 °C) and 2e' (19 mg, 0.075 mmol, 19%) as a white solid (Mp: 152–156 °C). ¹H NMR (300 MHz, CDCl₃) δ 7.71 (d, J=7.8 Hz, 2H, Ar), 7.34 (d, J=8.3 Hz, 2H, Ar), 4.91 (d, J=1.5 Hz, 1H, C=CHa), 4.85 (d, J=1.9 Hz, 1H, C=CHb), 3.96 (d, J=13.7 Hz, 1H, NCHaC=CH₂), 3.73 (d, J=14.2 Hz, 1H, NCHbC=CH₂), 3.58 (m, 1H, NCH₂CH), 2.72–2.67 (m, 2H, NCH₂CH), 2.44 (s, 3H, CH₃-Ar), 1.04 (d, J=5.9 Hz, 3H, CH₃); Anal. Calcd for C₁₃H₁₇NO₂S: C, 62.12; H, 6.82; N, 5.57. Found: C, 62.00; H, 6.93; N, 5.41.

3,4-Dimethyl-1-[(4-methylphenyl)sulfonyl]-2,5-dihydro-1*H***-pyrrole (2e').** ³¹ ¹H NMR (300 MHz, CDCl₃) δ 7.72 (d, J = 8.3 Hz, 2H, Ar), 7.31 (d, J = 8.0 Hz, 2H, Ar), 3.97 (s, 4H, 2 × NCH₂C), 2.42 (s, 3H, CH₃–Ar), 1.54 (s, 6H, 2 × CH₃); Anal. Calcd for C₁₃H₁₇NO₂S: C, 62.12; H, 6.82; N, 5.57. Found: C, 62.03; H, 6.65; N, 5.44.

tert-Butyl 3-methyl-4-methylenepyrrolidine-1-carboxylate (2f)²¹. Following general procedure for cycloisomerization, starting with 1f (100 mg, 0.398 mmol) after 1 h and purification through silica gel chromatography (Hex–AcOEt 20:1, $R_{\rm f}=0.36$ for 2f and $R_{\rm f}=0.67$ for 5f + 5f′, both in Hex–AcOEt 4:1) 2f (36 mg, 0.18 mmol, 36%) and 5f + 5f′ (70:30, 40 mg, 0.202 mmol, 40%) were obtained as colorless oils. Data for 2f: ¹H NMR (300 MHz, CDCl₃) δ 4.95 (bs, 1H, C=CHa), 4.89 (q, 1H, J=2.3 Hz, C=CHb), 3.98 (q, J=9.6 Hz, 2H, NCH₂C=CH₂), 3.78–3.61 (m, 1H, NCH₂CH), 2.91 (t, J=9.2 Hz, 1H, NCHaCH), 2.79–2.63 (m, 1H, NCHbCH), 1.47 (s, 9H, 3 × CH₃), 1.12 (d, J=6.7 Hz, 3H, CHCH₃); Anal. Calcd for C₁₁H₁₉NO₂: C, 66.97; H, 9.71; N, 7.10. Found: C, 67.05; H, 9.58; N, 7.01.

Data for (*E*)-tert-butyl allyl(prop-1-enyl)carbamate, (5f) and tert-butyl di(*E*)-prop-1-enylcarbamate, (5f'). 1 H NMR (300 MHz, CDCl₃) δ : 5f, 6.70 (d, J = 13.6 Hz, 1H, $CH = CHCH_3$), 5.75–5.62 (m, 1H, $CH_2CH = CH_2$), 5.06–4.99 (m, 2H, $CH_2CH = CH_2$), 4.83–4.69 (m, 1H, $CH = CHCH_3$), 4.00 (bs, 2H, $CH_2CH = CH_2$), 1.63 (dd, $J_1 = 6.7$ Hz, $J_2 = 1.6$ Hz, 3H,

CH=CH CH_3), 1.41 (s, 9H, C(CH_3)₃); **5f'**, 6.22 (d, J = 14.1 Hz, 2H, $2 \times CH = CHCH_3$), 5.31–5.19 (m, 2H, $2 \times CH = CHCH_3$), 1.60 (d, J = 6.6 Hz, 6H, 2 × CH=CHCH₃), 1.41 (s, 9H, $C(CH_3)_3$); ¹³C NMR (75 MHz, CDCl₃) δ 152.4, 133.1, 127.8. 127.0, 115.7, 115.2, 103.6, 81.0, 80.8, 46.5, 28.3, 15.4, 15.2.

1-(2,4-Dimethoxybenzyl)-3-methyl-4-methylidenepyrrolidin-2one (2g). Following general procedure for cycloisomerization, starting with 1g (100 mg, 0.383 mmol) after 16 h and purification through silica gel chromatography (Hex-AcOEt 4:1, $R_f = 0.60$ for 2g and $R_f = 0.55$ for 2g', both in Hex-AcOEt 1:1) 2g (45 mg, 0.17 mmol, 45%) and 2g' (12 mg, 0.045 mmol, 12%) were obtained as pale yellow oils. ¹H NMR (300 MHz, CDCl₃) δ 7.13 (d, J = 8.8 Hz, 1H, Ar), 6.46–6.44 (m, 2H, Ar), 5.03-5.00 (m, 2H, C=CH₂), 4.48 (s, 2H, CH₂Ar), 3.90-3.77 (m, 2H, $NCH_2C=CH_2$), 3.81 (s, 3H, OMe), 3.80 (s, 3H, OMe), 3.05 (q, J = 7.3 Hz, 1H, $CHCH_3$), 1.32 (d, J = 7.3 Hz, 3H, CHCH₃); 13 C NMR (75 MHz, CDCl₃) δ 175.7, 160.8, 159.0, 144.6, 131.2, 117.1, 108.0, 104.5, 98.7, 55.7, 55.7, 51.6, 41.8, 40.7, 15.8; IR (neat) 2924, 2851, 1711, 1649, 1615, 1589 cm⁻¹; Anal. Calcd for C₁₅H₁₉NO₃: C, 68.94; H, 7.33; N, 5.36. Found: C, 69.03; H, 7.24; N, 5.33.

1-(2,4-Dimethoxybenzyl)-4-methyl-3-methylidenepyrrolidin-2one (2g'). ¹H NMR (300 MHz, CDCl₃) δ 7.16 (d, J = 8.8 Hz, 1H, Ar), 6.47-6.43 (m, 2H, Ar), 6.00 (d, J = 2.4 Hz, 1H, C=CHa), 5.27 (d, J = 2.4 Hz, 1H, C=CHb), 4.56 (d, J =14.6 Hz, 1H, CHaAr), 4.48 (d, J = 14.6 Hz, 1H, CHbAr), 3.81 (s, 3H, OMe), 3.81 (s, 3H, OMe), 3.45 (t, J = 7.8 Hz, 1H, NCHaC=CH₂), 2.92-2.80 (m, 2H, NCHbC=CH₂ & NCH_2CH), 1.18 (d, J = 6.4 Hz, 3H, $CHCH_3$); IR (neat) 2921, 2858, 1645, 1589, 1507 cm⁻¹; Anal. Calcd for C₁₅H₁₉NO₃: C, 68.94; H, 7.33; N, 5.36. Found: C, 68.79; H, 7.40; N, 5.25.

(1E)-1-(prop-2-en-1-ylsulfonyl)prop-1-ene (5h).³² Following general procedure for cycloisomerization, starting with 1h (100 mg, 0.68 mmol) after 3 h was obtained after silica gel chromatography (Hex-AcOEt 4:1, $R_f = 0.43$ in Hex-AcOEt 2:1) **5h** (90 mg, 0.61 mmol, 90%) was obtained as colorless oil. ¹H NMR (300 MHz, CDCl₃) δ 6.96–6.82 (m, 1H, *CH*CH₃), 6.30 (apparent dd, $J_1 = 15.1$ Hz, $J_2 = 1.6$ Hz, 1H, SO_2CH =CH), 5.94–5.88 (m, 1H, CH₂CH=CH₂), 5.51 (d, J = 10.2 Hz, 1H, CH₂CH= CH_2), 5.44 (d, J = 17.0 Hz, 1H, $CH_2CH = CH_2$), 3.71 (AB system, 2H, $CH_2CH = CH_2$), 1.97 (dd, $J_1 = 6.7 \text{ Hz}, J_2 = 1.6 \text{ Hz}, 3H, CH = CHCH_3);$ ¹³C NMR (75 MHz, CDCl₃) δ 145.7, 128.6, 124.9, 124.4, 59.4, 17.4. IR (neat) 2975, 2922, 1640 cm⁻¹; Anal. Calcd for C₆H₁₀O₂S: C, 49.29; H, 6.89. Found: C, 49.38; H, 6.80.

1-(2,4-Dimethoxybenzyl)-3-ethyl-4-methylenepyrrolidin-2-one (9). Following general procedure for cycloisomerization, starting with 6 (100 mg, 0.68 mmol) after 16 h 9 was obtained after silica gel chromatography (Hex-AcOEt 6 : 1 to 2 : 1, $R_f = 0.23$ in Hex-AcOEt 2:1) (38 mg, 0.25 mmol, 38%) as a brown oil and 55 mg of a mixture of 7 + 8 which was separated by column chromatography (Hex-AcOEt 6:1) to give 15 mg of 7 (15%) and 35 mg of 8 (35%) as colorless oils. Data for 9: ¹H NMR (300 MHz, CDCl₃) δ 7.12 (d, J = 9.1 Hz, 1H, Ar), 6.46–6.42 (m, 2H, Ar), 5.05-5.01 (m, 2H, C=CH₂), 4.57 (d, J = 14.5 Hz, 1H, CHaAr), 4.39 (d, J = 14.6 Hz, 1H, CHbAr), 3.82–3.67 (m,

2H, NCH₂C=CH₂), 3.80 (s, 3H, OMe), 3.79 (s, 3H, OMe), 3.03 (bs, 1H, COCH), 1.93-1.72 (m, 2H, CHCH₂CH₃), 0.89 (t, J = 7.3 Hz, 3H, CHCH₂CH₃); ¹³C NMR (75 MHz, CDCl₃) δ 174.8, 160.6, 158.7, 142.1, 130.9, 116.9, 108.3, 104.3, 98.4, 55.4 (2 × C), 51.6, 47.7, 40.4, 24.0, 9.7; IR (neat) 2962, 2930, 2854, 1693, 1663, 1613, 1589 cm⁻¹; Anal. Calcd for C₁₆H₂₁NO₃: C, 69.79; H, 7.69; N, 5.09. Found: C, 69.89; H, 7.72; N, 5.03.

Data for (E)-N-allyl-N-(2,4-dimethoxybenzyl)but-2-enamide, (7). ¹H NMR (300 MHz, DMSO-d₆, 80 °C) δ 6.99 (d, J =8.4 Hz, 1H, Ar), 6.75–6.65 (m, 1H, CH= $CHCH_3$), 6.56 (d, J =2.3 Hz, 1H, Ar), 6.49 (dd, $J_1 = 8.4$ Hz, $J_2 = 2.3$ Hz, 1H, Ar), 6.36 (dd, $J_1 = 14.9$ Hz, $J_2 = 1.5$ Hz, 1H, $CH = CHCH_3$), 5.78-5.71 (m, 1H, $CH_2CH = CH_2$), 5.11-5.04 (m, 2H, $CH_2CH = CH_2$) 4.44 (s, 2H, ArCH₂N), 3.92 (AB system, 2H, $CH_2CH=CH_2$), 3.79 (s, 3H, OMe), 3.76 (s, 3H, OMe), 1.83 (dd, $J_1 = 6.8$ Hz, $J_2 = 1.6$ Hz, 3H, CH₃); ¹³C NMR (75 MHz, CDCl₃) 2 conformers δ 167.1, 160.4, 160.2, 158.0, 141.8, 141.8, 133.5, 133.4, 130.6, 127.9, 122.1, 117.5, 117.0, 116.3, 104.3, 103.9, 98.6, 98.3, 55.4, 55.2, 49.6, 48.2, 45.4, 43.2, 18.2;

Data for (E)-N-(2,4-dimethoxybenzyl)-N-((E)-prop-1-enyl)but-**2-enamide**, (8). ¹H NMR (300 MHz, DMSO-d₆, 80 °C) δ 6.96 (d, J = 13.6 Hz, 1H, NCH=CHCH₃), 6.83-6.75 (m, 1H, $CH = CHCH_3$), 6.77 (d, J = 10.9 Hz, 1H, $CH = CHCH_3$), 6.57 $(d, J = 2.4 \text{ Hz}, 1H, Ar), 6.47 (d, J = 8.4 \text{ Hz}, 1H, Ar), 6.49 (dd, J = 8.4 \text{ Hz}, 1H, Ar), 6.49 (dd, J = 8.4 \text{ Hz}, 1H, Ar), 6.49 (dd, J = 8.4 \text{ Hz}, 1H, Ar), 6.49 (dd, J = 8.4 \text{ Hz}, 1H, Ar), 6.49 (dd, J = 8.4 \text{ Hz}, 1H, Ar), 6.49 (dd, J = 8.4 \text{ Hz}, 1H, Ar), 6.49 (dd, J = 8.4 \text{ Hz}, 1H, Ar), 6.49 (dd, J = 8.4 \text{ Hz}, 1H, Ar), 6.49 (dd, J = 8.4 \text{ Hz}, 1H, Ar), 6.49 (dd, J = 8.4 \text{ Hz}, 1H, Ar), 6.49 (dd, J = 8.4 \text{ Hz}, 1H, Ar), 6.49 (dd, J = 8.4 \text{ Hz}, 1H, Ar), 6.49 (dd, J = 8.4 \text{ Hz}, 1H, Ar), 6.49 (dd, J = 8.4 \text{ Hz}, 1H, Ar), 6.49 (dd, J = 8.4 \text{ Hz}, 1H, Ar), 6.49 (dd, J = 8.4 \text{ Hz}, 1H, Ar), 6.40 (dd, J = 8.4 \text{$ $J_1 = 8.4$ Hz, $J_2 = 2.5$ Hz, 1H, Ar), 5.01–4.90 (m, 1H, NCH=CHCH₃), 4.68 (s, 2H, ArCH₂N), 3.82 (s, 3H, OMe), 3.75 (s, 3H, OMe), 1.85 (dd, $J_1 = 6.8$ Hz, $J_2 = 1.6$ Hz, 3H, CH₃), 1.62 (dd, $J_1 = 6.6$ Hz, $J_2 = 1.6$ Hz, 3H, CH₃); ¹³C NMR (75 MHz, CDCl₃) 2 conformers δ 169.8, 157.7, 157.3, 143.3, 129.3, 127.2, 126.8, 122.2, 117.7, 116.8, 110.6, 107.1, 104.1, 98.4, 55.4, 55.3, 44.2, 18.3, 15.5.

Diethyl 3,4-dimethylenecyclopentane-1,1-dicarboxylate (11a)³³. Following general procedure for cycloisomerization, starting with 10a (100 mg, 0.42 mmol) after 20 min 11a (80 mg, 0.34 mmol, 80%) was obtained after silica gel chromatography (Hex-AcOEt 20:1, $R_f = 0.42$ in Hex-AcOEt 6:1) as pale yellow oil. ¹H NMR (300 MHz, CDCl₃) δ 5.38 (s, 2H, 2 × C=CHa), 4.95 (s, 2H, $2 \times C$ =CHb), 4.19 (q, J = 7.2 Hz, 4H, 2 \times OCH₂CH₃), 3.03 (s, 4H, 2 \times CCH₂C), 1.24 (t, J = 7.0 Hz, 6H, $2 \times OCH_2CH_3$); Anal. Calcd for $C_{13}H_{18}NO_4$: C, 65.53; H, 7.61. Found: C, 65.42; H, 7.53.

(3,4-Dimethylenecyclopentane-1,1-diyl)bis(methylene)bis(oxy)bis(tert-butyldimethylsilane) (11b). Following general procedure for cycloisomerization, starting with 10b (100 mg, 0.26 mmol) after 20 min 11b (63 mg, 0.16 mmol, 63%) was obtained after silica gel chromatography (hexane, $R_f = 0.62$ in hexane) as colorless oil. ¹H NMR (300 MHz, CDCl₃) δ 5.34 (s, 2H, C= CH_2), 4.83 (s, 2H, C= CH_2), 3.39 (s, 4H, 2 × CH₂O), 2.26 (s, 4H, $2 \times CH_2$), 0.87 (s, 18H, $2 \times C(CH_3)_3$), 0.00 (s, 12H, $2 \times C(CH_3)_3$) $(CH_3)_2Si)$; ¹³C NMR (75 MHz, CDCl₃) δ 147.8, 104.6, 64.7, 47.0, 38.6, 25.9, 18.3, -5.5; IR (neat) 2955, 2930, 2857, 1676 cm^{-1} ; Anal. Calcd for $C_{21}H_{42}O_2Si_2$: C, 65.90; H, 11.06. Found: C, 65.73; H, 11.19.

General procedure for cyclotrimerization reaction

Catalyst [Ru]-II (7.2 mg, 0.008 mmol) was placed in a flamedried two-necked flask equipped with a condenser, and two cycles of vacuum-argon were performed. Anhydrous DMF (0.2 mL) was added and the suspension was heated at 120 °C for 12 min. The dark solution was cooled to r.t., anhydrous toluene (0.9 mL) was added followed by the addition of a mixture of diyne (0.42 mmol) and monoyne (1.27 mmol) in anhydrous toluene (1.5 mL). The reaction was gently refluxed until no more starting material was detected (TLC), cooled to r.t., filtered through Celite and solvents were removed under reduced pressure. The resulting dark-brown oil was purified by flash chromatography.

Diethyl 5-phenyl-1,3-dihydro-2*H*-indene-2,2-dicarboxylate (13a)³⁴. Following general procedure for cyclotrimerization, starting with 12a (100 mg, 0.42 mmol) after 24 h 13a (124.5 mg, 0.37 mmol, 87%) was obtained after silica gel chromatography (Hex-AcOEt 49:1 to 20:1, $R_f = 0.37$ in Hex-AcOEt 9:1) as a brown oil. ¹H NMR (300 MHz, CDCl₃) δ 7.56–7.53 (m, 2H, Ar), 7.42-7.37 (m, 4H, Ar), 7.33-7.23 (m, 2H, Ar), 4.21 (q, J =7.1 Hz, 4H, $2 \times OCH_2CH_3$), 3.64 (s, 2H, H-3), 3.63 (s, 2H, H-1), 1.26 (t, J = 7.1 Hz, 6H, $2 \times OCH_2CH_3$); Anal. Calcd for C₂₁H₂₂O₄: C, 74.54; H, 6.55. Found: C, 74.37; H, 6.63.

4,7-dimethyl-5-phenyl-1,3-dihydro-2*H*-indene-2,2dicarboxylate (13b). Following general procedure for cyclotrimerization, starting with 12b (100 mg, 0.259 mmol) after 24 h 13b was obtained after silica gel chromatography (Hex-AcOEt 20:1, $R_f = 0.38$ in Hex-AcOEt 9:1) (105.4 mg, 0.21 mmol, 83%) as a brown oil. ¹H NMR (300 MHz, CDCl₃) δ 7.40–7.22 (m, 16H, Ar), 5.15 (s, 4H, $2 \times ArCH_2O$), 3.61 (s, 4H, H-1 & H-3), 2.22 (s, 3H, CH₃-C4), 2.10 (s, 3H, CH₃-C7); ¹³C NMR (75 MHz, CDCl₃) δ 171.6, 142.0, 141.1, 139.3, 137.5, 135.5, 130.6, 130.0, 129.4, 128.6, 128.3, 128.0, 126.6, 67.4, 59.9, 40.3, 39.7, 18.6, 16.7; IR (neat) 3033, 2950, 2920, 1733, 1600 cm^{-1} ; Anal. Calcd for $C_{33}H_{30}O_4$: C, 80.79; H, 6.16. Found: C, 80.6 3; H, 6.39.

Dibenzyl 4,5,7-trimethyl-6-phenyl-1,3-dihydro-2*H*-indene-2,2dicarboxylate (13c). Following general procedure for cyclotrimerization, starting with 12b (100 mg, 0.259 mmol) after 24 h 13c was obtained after silica gel chromatography (Hex-AcOEt 20:1, $R_f = 0.44$ in Hex-AcOEt 9:1) (92.3 mg, 0.18 mmol, 71%) as a pale yellow oil. ¹H NMR (300 MHz, CDCl₃) δ 7.42–7.37 (m, 2H, Ar), 7.33–7.22 (m, 11H, Ar), 7.10–7.07 (m, 2H, Ar), 5.14 (s, 4H, 4H, $2 \times ArCH_2O$), 3.66 (s, 2H, H-3), 3.60 (s, 3H, H-1), 2.17 (s, 3H, CH₃), 1.90 (s, 3H, CH₃), 1.85 (s, 3H, CH₃); 13 C NMR (75 MHz, CDCl₃) δ 171.7, 142.0, 141.2, 137.4, 135.8, 135.5, 133.5, 129.5, 129.4, 128.9, 128.5, 128.3, 128.3, 128.0, 126.4, 67.4, 59.6, 40.4, 40.2, 17.4, 17.3, 16.4; IR (neat) 3063, 3033, 2924, 1733, 1601 cm⁻¹; Anal. Calcd for C₃₄H₃₂O₄: C, 80.93; H, 6.39. Found: C, 80.78; H, 6.37.

Notes and references

1 Reviews on non-metathetic behaviour of Ru catalysts: (a) B. Alcaide, P. Almendros and A. Luna, Chem. Rev., 2009, 109, 3819-3858; (b) B. Schmidt, Eur. J. Org. Chem., 2004, 1865–1880; (c) B. Alcaide and P. Almendros, Chem.-Eur. J., 2003, 9, 1258-1262.

- 2 For reviews on olefin metathesis, see: (a) S. P. Nolan and H. Clavier, Chem. Soc. Rev., 2010, 39, 3305-3316; (b) S. Monfette and D. E. Fogg, Chem. Rev., 2009, 109, 3783-3816; (c) T. J. Katz, Angew. Chem., Int. Ed., 2005, 44, 3010-3019; (d) R. Schrock, J. Mol. Catal. A: Chem., 2004, 213, 21-30; (e) A. H. Hoveyda and R. R. Schrock, Compr. Asymmetric. Catal., 2004, 1, 207-233; (f) R. H. Grubbs, Tetrahedron, 2004, 60, 7117-7140; (g) R. H. Grubbs and T. M. Trnka, in Ruthenium in Organic Synthesis, ed. S.-I. Murahashi, Wiley-VCH, Weinheim, 2004, Ch. 6.
- 3 A. Mallagaray, G. Domínguez, A. Gradillas and J. Pérez-Castells, Org. Lett., 2008, 10, 597-600.
- 4 S. Hanessian, S. Giroux and A. Larsson, Org. Lett., 2006, 8, 5481-5484.
- 5 (a) B. Schmidt, J. Org. Chem., 2004, 69, 7672-7687; (b) S. Fustero, M. Sánchez-Roselló, D. Jiménez, J. F. Sanz-Cervera, C. Pozo and J. L. Aceña, J. Org. Chem., 2006, 71, 2706-2714.
- 6 C. Menozzi, P. I. Dalko and J. Cossy, Synlett, 2005, 2449.
- 7 J. Gavenonis, R. V. Arroyo and M. L. Snapper, Chem. Commun., 2010, 46, 5692-5694, and references cited therein.
- 8 M. Arisawa, Y. Terada, K. Takahashi, M. Nakagawa and A. Nishida, J. Org. Chem., 2006, 71, 4255-4261.
- 9 (a) B. M. Trost, A. C. Gutierrez and E. M. Ferreira, J. Am. Chem. Soc., 2010, 132, 9206-9218; (b) B. M. Trost, A. C. Gutierrez and M. Ferreira, J. Am. Chem. Soc., 2008, 130, 16176–16177; (c) B. M. Trost and D. F. Toste, J. Am. Chem. Soc., 1999, 121, 9728–9729.
- 10 B. Çetinkaya, S. Demir, I. Özdemir, L. Toupet, D. Sémeril, C. Bruneau and P. H. Dixneuf, New J. Chem., 2001, 25, 519-521.
- 11 Y. Yamamoto, Y. Nakagai and K. Itoh, Chem.-Eur. J., 2004, 10, 231-236.
- 12 For cycloisomerization of enynes see: S. Kezuka, T. Okado, E. Niou and R. Takeuchi, Org. Lett., 2005, 7, 1711–1714, and ref. 6–11 cited therein.
- 13 Catalyst [Ru]-II (20 mg, 0.023 mmol) was loaded in an NMR tube and anhydrous DMF-d₇ (99.5 atom% D) (0.7 mL) and one drop of anhydrous DMF was added. The tube was filled with argon, closed and the temperature was progressively increased (30 °C min⁻¹) and ¹H was acquired every min.
- 14 For NMR study on [Ru]-II under variable temperatures, see: M. M. Gallagher, A. D. Rooney and J. J. Rooney, J. Organomet. Chem., 2008, 693, 1252-1260.
- 15 P. Serp, M. Hernandez, B. Richard and P. Kalck, Eur. J. Inorg. Chem., 2001, 2327-2336
- 16 A. J. Arduengo, R. Krafczyk and R. Schmutzler, Tetrahedron, 1999, 55, 14523-14534.
- 17 (a) S. Hyeok Hong, A. G. Wenzel, T. T. Salguero, M. W. Day and R. H. Grubbs, J. Am. Chem. Soc., 2007, 129, 7961-7968; (b) M. B. Herbert, Y. Lan, B. K. Keitz, P. Liu, K. Endo, M. W. Day, K. N. Houk and R. H. Grubbs, J. Am. Chem. Soc., 2012, 134, 7861-
- 18 A. Mallagaray, S. Medina, G. Domínguez and J. Pérez-Castells, Synlett, 2010, **14**, 2114–2118, and references cited therein.
- 19 D. Nečas, M. Turský, I. Tišlerová and M. Kotora, New J. Chem., 2006, **30**, 671–674.
- 20 S. Okamoto and T. Livinghouse, J. Am. Chem. Soc., 2000, 122, 1223-1224.
- 21 Y. Terada, M. Arisawa and A. Nishida, Angew. Chem., Int. Ed., 2004, 43, 4063-4067
- 22 D. A. Alonso, C. Nájera and M. Varea, Tetrahedron Lett., 2002, 43, 3459-3461.
- 23 B. L. Pagenkopf and T. Livinghouse, J. Am. Chem. Soc., 1996, 118, 2285-2286.
- 24 R. K. Singh, Synthesis, 1985, 54-55.
- 25 W. A. Nugent, J. Feldman and J. C. Calabrese, J. Am. Chem. Soc., 1995, 117, 8992-8998
- 26 T. S. Abram, R. Baker, C. M. Exon and V. B. Rao, J. Chem. Soc., Perkin Trans. 1, 1982, 285-294.
- 27 A. Fürstner, M. Liebl, C. W. Lehmann, M. Picquet, R. Kunz, C. Bruneau, D. Touchard and P. H. Dixneuf, Chem.-Eur. J., 2000, 6, 1847-1857.
- 28 S. Wolfe, S. Ro, C. Kim and Z. Shi, Can. J. Chem., 2001, 47, 1238-1258.
- 29 D. Crich, T. Hwang, S. Gastaldi, F. Recupero and D. J. Wink, J. Org. Chem., 1999, 64, 2877-2882.
- 30 D. Sémeril, C. Bruneau and P. H. Dixneuf, Helv. Chim. Acta, 2001, 84, 3335-3341
- 31 M. E. Krafft, L. V. R. Boñaga, J. A. C. Wright and C. Hirosawa, J. Org. Chem., 2002, 67, 1233-1246.
- 32 R. C. Fuson, C. C. Price and D. M. Burness, J. Org. Chem., 1946, 11, 475-481.
- 33 J. Y. Wu, B. N. Stanzl and T. Ritter, J. Am. Chem. Soc., 2010, 132, 13214–13216.
- 34 N. Saino, F. Amemiya, E. Tanabe, K. Kase and S. Okamoto, Org. Lett., 2006, 8, 1439-1442.